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Referências bibliográficas

- ALEXANDER, M.L. et al. *Appl. Surf. Sci.*, **255**, 127-129, 1998.
- ALFASSI, Z.B. (Ed.). **Non-destructive Elemental Analysis**. 1. ed. United Kingdom: Blackwell Science Ltd., London. Jeffries, T.E. Elemental analysis by laser ablation ICP-MS. Chapter 3, 2001.
- ARROWSMITH, P. *Anal. Chem.*, **59**, 1437, 1987.
- ASTON, F.W. *Philos. Mag.*, **38**, 707, 1919.
- BACHMANN, O.; DUNGAN, M. A.; BUSSY, F. Insights into shallow magmatic processes in large silicic magma bodies: trace element record in the Fish Canyon magma body, Colorado. *Contrib. Mineral. Petrol.*, **149** (3), 338-349, 2005.
- BECKER, J. S.; SAPRYKIN, A. I.; DIETZE, H.-J. *Int. J. Mass Spectrom., Ion Processes*, **164**, 81, 1997.
- BECKER, J. S.; DIETZE, H.-J. Determination of trace elements in geological samples by laser ablation inductively coupled plasma mass spectrometry. *Fresenius J. Anal. Chem.*, **365** (5), 429-434, 1999.
- BECKER, J. S.; DIETZE, H.-J. Inorganic mass spectrometric methods for trace, ultratrace, isotope, and surface analysis. *Int. J. Mass Spectrom.*, **197** (1-3), 1-35, 2000.
- BECKER, J. S.; PICKHARDT, C.; DIETZE, H.-J. Laser ablation inductively coupled plasma mass spectrometry for trace, ultratrace and isotope analyses of long-lived radionuclides in solid samples. *Int. J. Mass Spectrom.*, **202** (1), 283-297, 2000.
- BECKER, J. S. State-of-art and progress in precise and accurate isotope ratio measurements by ICP-MS and LA-ICP-MS. *J. Anal. At. Spectrom.*, **17** (9), 1172-1185, 2002a.
- BECKER, J. S. Applications of inductively coupled plasma mass spectrometry and laser ablation inductively coupled plasma mass spectrometry in materials science. *Spectrochim. Acta, Part B*, **57** (12), 1805-1820, 2002b.
- BECKER, J. S.; DIETZE, H.-J. State-of-the-art in inorganic mass spectrometry for analysis of high-purity materials. *Int. J. Mass Spectrom.*, **228** (2), 127-150, 2003.

BELLOTTO, V. R. **Estudos sobre a utilização de conchas de moluscos bivalves no monitoramento de poluição por metais, empregando a técnica de laser ablation ICP-MS.** Rio de Janeiro, agosto de 2000. 171p. Tese de Doutorado – Departamento de Química, Pontifícia Universidade Católica do Rio de Janeiro (PUC-Rio).

BELLOTTO, V. R.; MIEKELEY, N. Improvements in calibration procedures for the quantitative determination of trace elements in carbonate material (mussel shells) by laser ablation ICP-MS. *Fresenius J. Anal. Chem.*, **367** (7) 635-640, 2000.

BLEINER, D.; GÜNTHER, D. Theoretical description and experimental observation of aerosol transport processes in laser ablation inductively coupled plasma mass spectrometry. *J. Anal. At. Spectrom.*, **16** (5), 449-456, 2001.

BORMAN, S.; RUSSELL, H.; SIUZDAK, G. A Mass Spectrometry Timeline. **Today's Chemist**, 47-49, 2003.

BRECH, F.; CROSS, L. *Appl. Spectrosc.*, **16**, 59, 1962.

BRUHN, C. H. L. How much oil and gas from deepwater? The experience of Brazil. Abstract from **ASPO's 2005 - IV International Workshop on Oil and Gas Depletion**, 19-20 may 2005, Lisbon, Portugal.

BURNS, R. G. **Mineralogical applications of crystal field theory.** Cambridge University Press, Cambridge, 1970. 224p.

CHENERY, S.; Cook, J. M. *J. Anal. At. Spectrom.*, **8**, 299-303, 1993.

CHRISTOPHER, C. et al. Determination of elements in National Bureau of Standards' geological standard reference materials by neutron activation analysis. *Anal. Chem.*, **54**, 1623-1627, 1982.

CORNELL, E. A. et al. Observation of Bose-Einstein condensation in a dilute atomic vapor. *Science*, **269**, 198, 1995.

CROCK, J. G.; LICHTE, F. E.; BRIGGS, P. H. Determination of elements in National Bureau of Standards' geological reference materials SRM 278 obsidian and SRM 688 basalt by inductively coupled argon plasma – atomic emission spectrometry. *Geost. Newslett.*, **7** (2), 335-340, 1983.

DARKE, S. A.; Tyson, J. F. *J. Anal. At. Spectrom.*, **8** (2), 145-209, 1993.

DE LAETER, J. R. **Applications of Inorganic Mass Spectrometry**, Wiley-Interscience Series on Mass Spectrometry, John-Wiley & Sons, Inc., USA, 2001. 474p.

DEMPSTER, A. J. *Phys. Rev.*, **11**, 316, 1918.

DULSKI, T. R. **Trace elemental analysis of metals: methods and techniques.** 1st Edition, Marcel Dekker, Inc., USA, 1999. 581p.

DURRANT, S. F. Laser ablation inductively coupled plasma mass spectrometry: achievements, problems, prospects. *J. Anal. At. Spectrom.*, **14**, 1385-1403, 1999.

EASTMAN, T. E. Internet, site: <http://www.plasmas.org/index.html>, Plasmas International, 1999.

EBDO, L.; Foulkes, M.; Sutton, K. *J. Anal. At. Spectrom.*, **12**, 213, 1997.

EGGINS, S.M. Laser ablation ICP-MS analysis of geological materials prepared as lithium borate glasses. *Geost. Newslett.*, **27** (2), 147-162, 2003.

EVENSEN, N. M.; HAMILTON, P. J.; O'NIONS, R. K. Rare earth abundances in chondritic meteorites, *Geochim. Cosmochim. Acta*, **42**, 1199-1212, 1978.

FASSEL, V. A. *Science*, **202**, 185, 1978.

GEERTSEN, C. et al. *J. Anal. At. Spectrom.*, **9**, 17, 1994.

GOLDSTEIN, E. *Berl. Ber.*, **39**, 691, 1886.

GOVINDARAJU, K. (Editor-in-Chief). 1989 Compilation of working values and sample description for 272 geostandards. *Geost. Newslett.*, **13**, Special Issue, jul. 1989.

GRAY, A. L. Solid sample introduction by laser for inductively coupled plasma source mass spectrometry. *Analyst*, **110**, 551-556, 1985.

GREENFIELD, S.; I. Jones; Berry, C. T. High-pressure plasmas as spectroscopic emission sources. *Analyst*, **89**, 713-720, 1964.

GRÉGOIRE, D. C. et al. *Chem. Geol.*, **124**, 91, 1995.

GÜNTHER, D. et al. Elemental analyses using laser ablation – inductively coupled plasma – mass spectrometry (LA-ICP-MS) of geological samples fused with Li₂B₄O₇ and calibrated without matrix-matched standards. *Mikrochim. Acta*, **136** (3-4), 101-107, 2001.

GÜNTHER, D.; JACKSON, S. E.; LONGERICH, H. P. Laser ablation and arc/spark solid sample introduction into inductively coupled plasma mass spectrometers. *Spectrochim. Acta, Part B*, **54** (3-4), 381-409, 1999.

GÜNTHER, D.; HEINRICH, C. A. Comparison of the ablation behaviour of 266 nm Nd:YAG and 193 nm ArF excimer lasers for LA-ICP-MS analysis. *J. Anal. At. Spectrom.*, **14** (9), 1369-1374, 1999.

GÜNTHER, D.; HORN, I.; HATTENDORF, B. Recent trends and developments in laser ablation-ICP-mass spectrometry. *Fresenius J. Anal. Chem.*, **368**, 4-14, 2000.

GÜNTHER, D.; HATTENDORF, B. Solid sample analysis using laser ablation inductively coupled plasma mass spectrometry. *Trends in Analytical Chemistry*, **24** (3), 255-265, 2005.

HEMMERLIN, M.; MERMET, J.-M. Determination of elements in polymers by laser ablation inductively coupled plasma atomic emission spectrometry: effect of the laser beam wavelength, energy and masking on the ablation threshold and efficiency. *Spectrochim. Acta, Part B*, **51**, 579-589, 1996.

HOLLIMAN, C.L.; REMPEL, D.L.M.; CROSS, M.L. *Mass Spectrom. Rev.*, **13**, 105, 1994.

HOLLOCHER, K.; FAKHRY, A.; RUIZ, J. Trace element determinations for USGS basalt BHVO-1 and NIST standard reference materials 278, 688 and 694 by inductively coupled plasma – mass spectrometry. *Geost. Newslett.*, **19** (1), 5-40, 1995.

HOUK, R. S. et al. Inductively coupled argon plasmas as ion sources for mass spectrometric determination of trace elements. *Anal. Chem.*, **52**, 2283-2289, 1980.

JEFFRIES, T. E.; PERKINS, W. T.; PEARCE, N. J. G. Comparisons of infrared and ultraviolet laser probe microanalysis inductively coupled plasma mass spectrometry in mineral analysis. *Analyst*, **120** (5), 1365-1371, 1995.

JEFFRIES, T. E.; JACKSON, S. E.; LONGERICH, H. P. *J. Anal. At. Spectrom.*, **13**, 935-940, 1998.

JOCHUM, K. P.; SEUFERT, M.; BEST, S. *Fresenius J. Anal. Chem.*, **309**, 308, 1981.

JOCHUM, K. P. *Spectrosc. Eur.*, **9**, 22, 1997.

JOCHUM, K. P. et al. *Fresenius J. Anal. Chem.*, **359**, 385, 1997.

JOHNSON, E. G.; NIER, A. O. *Phys. Rev.*, **91**, 10, 1953.

KLEMM, W.; BOMBACH, G. A simple method of target preparation for bulk analysis of powder samples by laser ablation inductively coupled plasma mass spectrometry (LA-ICP-MS), *Fresenius J. Anal. Chem.*, **370** (5), 641-646, 2001.

LICHTE, F. E. Determination of elemental content of rocks by laser ablation inductively coupled plasma mass spectrometry. *Analytical Chemistry*, **67** (14), 2479-2485, 1995.

LONGERICH, H. P.; JACKSON, S. E.; GÜNTHER, D. Laser ablation inductively coupled plasma mass spectrometric transient signal data acquisition and analyte concentration calculation. *J. Anal. At. Spectrom.*, **11** (9), 899-904, 1996.

LONGERICH, H. P.; GÜNTHER, D.; JACKSON, S. E. Elemental fractionation in laser ablation inductively coupled plasma mass spectrometry. *Fresenius J. Anal. Chem.*, **355** (5-6), 538-542, 1996.

MARGULIS, W. Nota Didática, de aula ministrada no Departamento de Física da PUC-Rio, Rio de Janeiro – Brasil, março 1992.

MATTAUCH, J. *Phys. Rev.*, **50**, 617 e 1089, 1936.

MAY, T. W.; WIEDMEYER, R. H. A Table of Polyatomic Interferences in ICP-MS. *Atomic Spectroscopy*, **19** (5), 150-155, 1998.

MIEKELEY, N.; AMATO, M. O. Fast mercury determination in biological samples by ICP-MS using “Minitube Furnace Catalytic Combustion (MFCC)”. *Atomic Spectroscopy*, **18** (6), 186-191, 1997.

MILLER, J. C.; MILLER, J. N. **Statistics for Analytical Chemistry**. p. 104-115, 3rd edition, Ellis Horwood Limited, Chichester, England, 1993, 233p.

MONTASER, A.; FASSEL, V. A. Inductively coupled plasma as an atomization cell for atomic fluorescence spectrometry. *Anal. Chem.*, **48**, 1490-1499, 1976.

MONTASER, A. (Ed.). **Inductively Coupled Plasma Mass Spectrometry**. 1. ed. Wiley-VCH, Inc., New York USA, 1998, 964p.

NIER, A. O. *Nat. Bur. Stand. Circ. (U.S.)*, **522**, 29-36, 1953.

NIST (ou NBS). National Bureau of Standards, Certificate of Analysis of Standard Reference Material 688, basalt rock, aug. 1981.

NIST. National Institute of Standards & Technology, Certificate of Analysis of Standard Reference Material 278, obsidian rock, mar. 1992.

ODEGARD, M.; HAMESTER, M. Preliminary investigation into the use of a high resolution inductively coupled plasma – mass spectrometer with laser ablation for bulk analysis of geological materials fused with Li₂B₄O₇. *Geost. Newslett.*, **21** (2), 245-252, 1997.

ORIHASHI, Y.; HIRATA, T. Rapid quantitative analysis of Y and REE abundances in XRF glass bead for selected GSJ reference rock standards using Nd:YAG 266 nm UV laser ablation ICP-MS, *Geochemical Journal*, **37** (3), 401-412, 2003.

PAUL, W.; STEINWEDEL, H. S. Z. *Naturforsch.*, **8a**, 448-450, 1953.

PEREIRA, C. E. de B.; MIEKELEY, N.; POUPEAU, G.; KÜCHLER, I. L. Determination of minor and trace elements in obsidian rock samples and archaeological artifacts by laser ablation inductively coupled plasma mass spectrometry using synthetic obsidian standards. *Spectrochim. Acta, Part B*, **56**, 1927-1940, 2001.

PEREIRA, C. E. de B. **Caracterização química de vidros vulcânicos por laser ablation ICP-MS e outras técnicas visando estudos de proveniência de artefatos arqueológicos**. Rio de Janeiro, dezembro de 2000. 294p. Tese de Doutorado – Departamento de Química, Pontifícia Universidade Católica do Rio de Janeiro (PUC-Rio).

PERKINS, W. T.; PEARCE, N. J. G.; JEFFRIES, T. E. Laser ablation inductively coupled plasma mass spectrometry: a new technique for the determination of trace and ultra-trace elements in silicates, *Geochimica et Cosmochimica Acta*, **57** (2), 475-482, 1993.

PHILP, R. P. Formation and Geochemistry of Oil and Gas, Chapter 9, volume 7: Sediments, Diagenesis and Sedimentary Rocks, in **Treatise on Geochemistry**, Executive Editors: Holland, H. D.; Turekian, K. K., 1st Edition, Elsevier Ltd., Oxford, United Kingdom, 2004.

PICKHARDT, C.; BECKER, J. S.; DIETZE, H.-J. A new strategy of solution calibration in laser ablation inductively coupled plasma mass spectrometry for multielement trace analysis of geological samples, *Fresenius J. Anal. Chem.*, **368** (2-3), 173-181, 2000.

RAUT, N. M. et al. Determination of lanthanides in rock samples by inductively coupled plasma mass spectrometry using thorium as oxide and hydroxide correction standard. *Spectrochim. Acta, Part B*, **58** (5), 809-822, 2003.

REED, T. B. Induction-Coupled Plasma Torch. *J. Appl. Phys.*, **32**, 821-824, 1961.

REGAL, C. A.; GREINER, M.; JIN, D. S. Observation of resonance condensation of fermionic atom pairs. *Phys. Rev. Lett.*, **92** (4), 1-4, 2004.

ROCHA, C. L. **Análise de fronteiras de reservatório de petróleo através de geoquímica de superfície e mineração de dados**. Tese de Mestrado, 143p. Engenharia Civil da COPPE / Universidade Federal do Rio de Janeiro, 2005.

ROLLINSIN, H. R. **Using Geochemical Data: Evaluation, Presentation, Interpretation**, Wiley, New York – USA, 1993. 352p.

RUSAK, D. A. et al. *Trends in Analytical Chemistry*, **17** (8-9), 453-461, 1998.

RUSSO, R. E.; MAO, X.; MAO, S. S. The physics of laser ablation in microchemical analysis. *Anal. Chem.*, **74** (3), 71A-77A, 2002.

RUSSO, R. E. et al. Laser ablation in analytical chemistry – a review. *Talanta*, **57** (3), 425-451, 2002a.

RUSSO, R. E. et al. Femtosecond laser ablation ICP-MS. *J. Anal. At. Spectrom.*, **17** (9), 1072-1075, 2002b.

SAPRYKIN, A. I.; BECKER, J. S.; DIETZE, H.-J. *Fresenius J. Anal. Chem.*, **364**, 763, 1999.

SCHELLES, W. et al. *Appl. Spectrosc.*, **49**, 939, 1995.

SCHELLES, W. et al. *Fresenius J. Anal. Chem.*, **355**, 858, 1996.

SKOOG, D. A.; HOLLER, F. J.; NIEMAN, T. A. **Principles of Instrumental Analysis**. 5th Edition by Thomson Learning, Inc., USA, 1998. 849p.

STEPHENS, W. E. *Phys. Rev.*, **69**, 691, 1946.

TANNER, M.; GÜNTHER, D. In-torch laser ablation sampling for inductively coupled plasma mass spectrometry. *J. Anal. At. Spectrom.*, **20** (9), 987, 2005.

TANNER, S. D.; BARANOV, V. I. Theory, Design, and Operation of a Dynamic Reaction Cell for ICP-MS. *Atomic Spectroscopy*, **20** (2), 45-52, 1999.

TAYLOR, H. E. **Inductively Coupled Plasma - Mass Spectrometry: Practices and Techniques**, Academic Press, USA, 2001, 294p.

THOMPSON, M.; GOULTER, J. E.; SIEPER, F. Laser ablation for the introduction of solid samples into an inductively coupled plasma for atomic emission spectrometry. *Analyst*, **106**, 32-39, 1981.

THOMSON, J.J. *Phil. Mag.*, **13**, 561, 1907.

THOMSON, J.J. *Phil. Mag.*, **6**, 752-767, 1911.

VALCÁRCEL, M. **Principles of Analytical Chemistry - A Textbook**, edited by Springer-Verlag, Heidelberg, Germany, 2000, 371p.

VAN HECKE, G. R.; KARUKSTIS, K. K. **A Guide to Lasers in Chemistry**. Edition by Jones and Bartlett Publishers, Inc., USA, 1998. 252p.

VAN HEUZEN, A. A. Analysis of solids by laser ablation-inductively coupled plasma-mass spectrometry (LA-ICP-MS) - I. matching with a glass matrix. *Spectrochim. Acta, Part B*, **46** (14), 1803-1817, 1991.

VENZAGO, C. et al. *J. Anal. At. Spectrom.*, **13**, 189, 1998.

WEDEPOHL, K. H. The composition of the continental crust. *Geochim. et Cosmochim. Acta*, **59** (7), 1217-1232, 1995.

WILLIAMS, J. G.; JARVIS, K. E. Preliminary assessment of laser ablation inductively coupled plasma mass spectrometry for quantitative multi-element determination in silicates. *J. Anal. At. Spectrom.*, **8** (1), 25-34, 1993.

YOST, R. A.; ENKE, C. G. *J. Am. Chem. Soc.*, **100**, 2274-2275, 1978.

YU, Z. S.; NORMAN, M. D.; ROBINSON, P. Major and trace element analysis of silicate rocks by XRF and laser ablation ICP-MS using lithium borate fused glasses: matrix effects, instrument response and results for international reference materials, *Geostand. Newslet.*, **27** (1), 67-89, 2003.

ANEXOS

Tabela A.1 (Anexo I) – Figuras de mérito das curvas analíticas nas análises por SN-ICPMS para validação dos padrões de vidro borato. Valores para os coeficientes de determinação (R^2) e da inclinação (b) das retas contendo sete pontos, e que passam pelo ponto de origem.

Analito	m/z	R^2	b (kg/ μ g)	faixa dinâmica (μ g/kg)
Na	23	0,9993	0,0326	0-1100
Al	27	0,9574	0,0316	0-100
K	39	0,9995	0,0396	0-1100
Ca	44	0,8616	0,0216	0-1100
Sc	45	0,9981	0,0521	0-100
Ti	47	0,8212	0,0071	0-100
Ti	48	0,6072	0,0094	0-100
V	51	0,9993	0,039	0-100
Cr	53	0,9997	0,0041	0-100
Mn	55	0,9992	0,041	0-100
Fe	57	0,9911	0,0028	0-1100
Co	59	0,9996	0,031	0-100
Ni	60	0,9990	0,0076	0-100
Ni	62	0,9988	0,0012	0-100
Zn	66	0,9996	0,0048	0-100
Ga	71	0,9996	0,018	0-100
Rb	85	0,9998	0,045	0-100
Sr	88	0,9995	0,054	0-100
Y	89	0,9988	0,068	0-100
Zr	90	0,9997	0,033	0-100
Nb	93	0,9998	0,058	0-100
Rh	103	0,8372	3488	0-100
In	115	0,9959	0,176	0-100
Ba	138	0,9997	0,051	0-100
La	139	0,9996	0,072	0-100
Ce	140	0,9998	0,062	0-100
Pr	141	0,9997	0,074	0-100
Nd	142	0,9997	0,021	0-100
Sm	152	0,9997	0,020	0-100
Eu	153	0,9998	0,038	0-100
Gd	158	0,9999	0,018	0-100
Tb	159	0,9997	0,067	0-100
Dy	164	0,9999	0,019	0-100
Ho	165	0,9998	0,063	0-100
Er	166	0,9998	0,022	0-100
Tm	169	0,9997	0,062	0-100
Yb	174	0,9996	0,020	0-100
Lu	175	0,9995	0,057	0-100
Ta	181	0,9997	0,054	0-100
Pb	208	0,9994	0,021	0-100
Th	232	0,9999	0,042	0-100
U	238	0,9998	0,039	0-100

Anexo II

Tabela A.2 – Resultados da validação de alvo padrão: vidro borato (1:5) de basalto NIST SRM 688. Fragmentos dos alvos de calibração foram dissolvidos em HNO₃ e as soluções analisadas por SN-ICPMS.

Elemento	Concentração ($\mu\text{g kg}^{-1}$)	
	Valor esperado	SN-ICPMS
Ce	2.000	2.300 ± 110
Co	8.580	7.860 ± 390
Cr	55.300	55.400 ± 3.700
Er	316	410 ± 30
Ga	2.832	3.260 ± 200
Gd	483	610 ± 30
Ho	110	130 ± 20
La	816	1.010 ± 50
Mn	215.000	290.000 ± 22.000
Nd	1.399	1.620 ± 220
Pr	333	330 ± 20
Sr	27.000	36.600 ± 1.800
Ta	52	50 ± 10
U	62	50 ± 10
Y	2.815	3.500 ± 200
Zn	13.500	14.600 ± 1.500

Tabela A.3 – Resultados da validação de alvo padrão: vidro borato (1:11) de basalto NIST SRM 688. Fragmentos dos alvos de calibração foram dissolvidos em HNO₃ e as soluções analisadas por SN-ICPMS.

Elemento	Concentração ($\mu\text{g kg}^{-1}$)	
	Valor esperado	SN-ICPMS
Ce	1.004	1.030 ± 30
Co	4.310	3.350 ± 280
Cr	27.800	24.850 ± 930
Er	159	159 ± 10
Eu	74	66 ± 3
Ga	1.422	1.430 ± 80
Gd	243	253 ± 20
Ho	55	47 ± 5
La	410	430 ± 30
Lu	22	12 ± 1
Nb	435	280 ± 20
Nd	703	670 ± 60
Pr	167	130 ± 10
Y	1.413	1.500 ± 100
Zn	6.775	6.500 ± 200

Tabela A.4 – Resultados da validação de alvo padrão: vidro borato (1:11) de obsidiana NIST SRM 278. Fragmentos dos alvos de calibração foram dissolvidos em HNO₃ e as soluções analisadas por SN-ICPMS.

Elemento	Concentração ($\mu\text{g kg}^{-1}$)	
	Valor esperado	SN-ICPMS
Ce	5.457	4.400 \pm 240
Cr	532	800 \pm 250
Er	344	310 \pm 20
Ga	1.847	1.500 \pm 130
Gd	479	530 \pm 60
Ho	110	90 \pm 10
La	2.519	2.300 \pm 130
Lu	54	40 \pm 10
Mn	34.000	37.000 \pm 3.000
Nb	1.797	1.140 \pm 60
Nd	2.267	2.100 \pm 160
Sm	470	420 \pm 40
Sr	5.037	5.200 \pm 740
Ta	126	100 \pm 10
Th	1.016	930 \pm 120
U	395	340 \pm 20
Zn	4.450	3.900 \pm 300

Tabela A.5 – Resultados da validação de alvo padrão: vidro borato (1:23) de obsidiana NIST SRM 278. Fragmentos dos alvos de calibração foram dissolvidos em HNO₃ e as soluções analisadas por SN-ICPMS.

Elemento	Concentração ($\mu\text{g kg}^{-1}$)	
	Valor esperado	SN-ICPMS
Ce	2.697	1.440 \pm 130
Co	830	624 \pm 520
Cr	263	n.a.
Eu	28	5 \pm 1
Ga	913	440 \pm 20
Gd	236	130 \pm 10
Ho	54	16 \pm 2
La	1.245	700 \pm 70
Lu	27	20 \pm 10
Mn	17.000	22.000 \pm 9.000
Nb	888	530 \pm 200
Nd	1.120	1.200 \pm 350
Pr	290	320 \pm 140
Sm	232	250 \pm 120
Sr	2.490	3.000 \pm 1.100
Ta	62	50 \pm 30
Th	502	560 \pm 250
Tm	26	15 \pm 10
U	195	210 \pm 90
Y	1.493	1.700 \pm 700
Zn	2.200	2.500 \pm 1.400

Tabela A.6 – Resultados da validação de alvo padrão: vidro borato (1:47) de obsidiana NIST SRM 278. Fragmentos dos alvos de calibração foram dissolvidos em HNO₃ e as soluções analisadas por SN-ICPMS.

Elemento	Concentração ($\mu\text{g kg}^{-1}$)	
	Valor esperado	SN-ICPMS
Ce	1.346	3.000 ± 1.300
Cr	131	n.a.
Er	85	185 ± 90
Ga	456	440 ± 20
Gd	118	130 ± 10
Ho	27	50 ± 30
La	621	1.500 ± 550
Mn	8.000	9.000 ± 800
Nd	559	600 ± 60
Pr	145	150 ± 10
Sm	116	110 ± 10
Sr	1.243	1.560 ± 130
Th	251	260 ± 30
Tm	105	90 ± 10
U	97	100 ± 10
Y	746	850 ± 50
Zn	1.100	1.400 ± 200

Anexo III

Tabela A.7 – Resultados da determinação de elementos no vidro borato (1:5) de basalto GSJ JB-2.

Elemento	Concentração (mg kg^{-1}) ^a	
	Valor de referência	Este trabalho
Sc	54	59 ± 1
P	440	330 ± 15
V	578	580 ± 6
Cr	27	26 ± 31^b
Mn	1.550	1.760 ± 60
Fe	10 %	(11 ± 1) %
Co	40	44 ± 8
Ni	14	14 ± 3^b
Zn	110	110 ± 10
Ga	17	14 ± 1
Rb	6,2	$6,5 \pm 1,7$
Sr	178	172 ± 5
Y	25	25 ± 1
Zr	51	55 ± 6
Sb	0,270	$0,310 \pm 0,005^b$
Ba	208	226 ± 17
La	2,4	$2,4 \pm 0,4$
Ce	6,8	$6,8 \pm 0,5$
Nd	6,7	$7,0 \pm 0,7$
Sm	2,3	$2,5 \pm 0,2^b$
Eu	0,86	$0,80 \pm 0,07^b$
Tm	0,45	$0,38 \pm 0,04$

^a Unidades diferentes de concentração são indicadas.

^b Exatidão e repetitividade da concentração melhoram com o emprego de câmara ciclônica

Tabela A.8 – Resultados da determinação de elementos no vidro borato (1:5) de granito NIM-G SARM-1.

Elemento	Concentração (mg kg^{-1})	
	Valor de referência	Este trabalho
Na	24.930	23.270 ± 250
Al	63.930	64.600 ± 700
Si	353.860	353.500 ± 11.600
K	41.420	39.930 ± 640
Ca	5.570	5.300 ± 1.030
Zn	50	52 ± 2
Ga	27,0	$32,8 \pm 2,0$
Rb	325	274 ± 5
Zr	300	$295 \pm 12^{\text{a}}$
Nb	53	72 ± 7
Ba	120	129 ± 17
La	109	$120 \pm 6^{\text{a}}$
Ce	195	191 ± 12
Pr	19	$23 \pm 1^{\text{a}}$
Nd	72	$66 \pm 2^{\text{a}}$
Sm	16	$14 \pm 1^{\text{a}}$
Eu	0,35	$0,31 \pm 0,05^{\text{a}}$
Gd	14,0	$12,8 \pm 1,4^{\text{a}}$
Tb	3,0	$2,6 \pm 0,4^{\text{a}}$
Dy	17	$19 \pm 4^{\text{a}}$
Ho	3,6	$4,4 \pm 0,4$
Er	10,5	$11,1 \pm 1,4^{\text{a}}$
Tm	2,0	$2,2 \pm 0,3^{\text{a}}$
Yb	14,2	$11,9 \pm 0,8^{\text{a}}$
Lu	2,00	$1,96 \pm 0,12^{\text{a}}$
Hf	12,4	$13,1 \pm 2,5^{\text{a}}$
Ta	0,20	$0,20 \pm 0,03^{\text{a}}$
Th	51	54 ± 3
U	15	15 ± 2

^a Exatidão e repetitividade da concentração melhoram com o emprego de câmara ciclônica

Anexo IV

Tabela A.9 – Resultados da determinação de elementos no vidro borato (1:5) de basalto (USGS, BHVO-2).

Elemento	Concentração (mg kg^{-1})	
	Valor de referência	Este trabalho
Al	71.600	123.000 ± 15.000
Ba	130	170 ± 30
Ca	81.700	79.000 ± 7.000
Ce	38	40 ± 6
Co	45	43 ± 2
Cr	280	270 ± 5
Fe	86.300	81.000 ± 2.800
La	15	21 ± 1
Mn	1.290	1.290 ± 80
Nd	25	31 ± 2
Ni	119	90 ± 20
P	1.200	820 ± 60
Rb	9,8	8 ± 2
Sc	32	32 ± 2
Si	233.000	480.000 ± 47.000
Sr	389	350 ± 30
V	317	327 ± 6
Y	26	28 ± 2
Zn	103	110 ± 2
Zr	172	247 ± 6

Tabela A.10 – Resultados da determinação de elementos no vidro borato (1:5) de basalto (USGS, BCR-2).

Elemento	Concentração (mg kg^{-1})	
	Valor de referência	Este trabalho
Al	71.400	65.000 ± 7.000
Ba	683	760 ± 40
Ce	53	54 ± 7
Co	37	38 ± 2
Eu	2,0	$1,8 \pm 0,4$
Fe	96.500	79.000 ± 2.700
Ga	23	19 ± 2
Ho	1,3	$1,0 \pm 0,1$
K	14.900	13.600 ± 600
La	25	26 ± 1
Mn	1.520	1.300 ± 80
Nd	28	37 ± 3
Ni	119	90 ± 20
P	1.500	1.100 ± 80
Rb	48	40 ± 2
Sc	33	27 ± 2
Si	253.000	278.000 ± 28.000
Sr	346	280 ± 20
V	416	424 ± 8
Y	37	36 ± 3
Zn	127	130 ± 20
Zr	188	194 ± 5

Tabela A.11 – Resultados da determinação de elementos no vidro borato (1:5) de basalto (USGS, BIR-1).

Elemento	Concentração (mg kg ⁻¹)	
	Valor de referência	Este trabalho
Al	82.035	95.000 ± 11.000
Ca	95.050	64.000 ± 6.000
Ce	1,9	2 ± 1
Co	52	57 ± 3
Cr	370	490 ± 10
Fe	79.040	80.900 ± 2.700
Mn	1.350	1.290 ± 80
Ni	170	140 ± 20
Sc	44	36 ± 2
Si	224.200	263.000 ± 30.000
Sr	110	100 ± 10
V	310	350 ± 6
Y	16	20 ± 2
Zn	70	80 ± 10

Tabela A.12 – Resultados da determinação de elementos no vidro borato (1:5) de folhelho (USGS, SDO-1).

Elemento	Concentração (mg kg ⁻¹)	
	Valor de referência	Este trabalho
Al	64.940	99.000 ± 12.000
Ce	79	110 ± 10
Eu	1,6	1,8 ± 0,4
Fe	65.300	88.000 ± 3.000
Ga	17	21 ± 3
Ho	1,2	0,7 ± 0,1
Mn	325	440 ± 50
Ni	99	90 ± 20
P	480	640 ± 50
Rb	48	40 ± 2
Sc	13	14 ± 1
Si	230.400	420.000 ± 40.000
Sr	75	90 ± 10
Tb	1,2	1,5 ± 0,1
V	160	247 ± 5

Tabela A.13 – Resultados da determinação de elementos no vidro borato (1:5) de folhelho (USGS, SGR-1).

Elemento	Concentração (mg kg^{-1})	
	Valor de referência	Este trabalho
Al	34.510	90.000 ± 10.000
Ca	59.900	71.000 ± 6.000
Fe	21.200	43.000 ± 1.700
Gd	2,0	$2,6 \pm 0,2$
Mn	267	490 ± 50
Sb	3,4	$4,9 \pm 0,4$
Si	132.000	440.000 ± 43.000
V	130	310 ± 10

Tabela A.14 – Resultados da determinação de elementos no vidro borato (1:5) de folhelho (USGS, SCo-1).

Elemento	Concentração (mg kg^{-1})	
	Valor de referência	Este trabalho
Al	72.500	60.000 ± 7.000
Ba	570	480 ± 30
Ce	62	50 ± 10
Co	11	7 ± 1
Fe	35.900	22.000 ± 1.500
Ga	15	14 ± 2
K	23.000	21.000 ± 1.000
La	30	24 ± 1
Mn	410	240 ± 50
Nd	26	29 ± 2
Pb	31	20 ± 5
Pr	6,6	$5,5 \pm 1,0$
Rb	110	99 ± 3
Sb	2,5	$3,6 \pm 0,3$
Sc	11	6 ± 1
Si	293.600	145.000 ± 24.000
Sr	170	100 ± 10
V	130	86 ± 3
Y	26	16 ± 2
Zn	100	90 ± 10
Zr	160	123 ± 4