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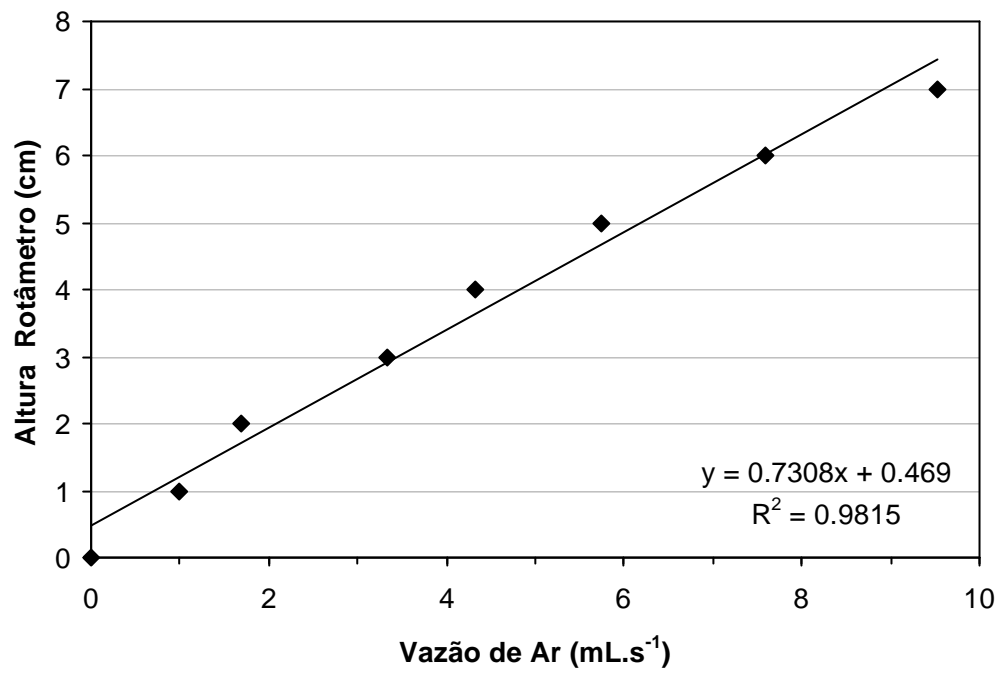
## 10 ANEXOS

### 10.1.

**Aparelho Zeta Master- Malvern Instruments, equipado com um microprocessador empregado para a Medição do Potencial Zeta**

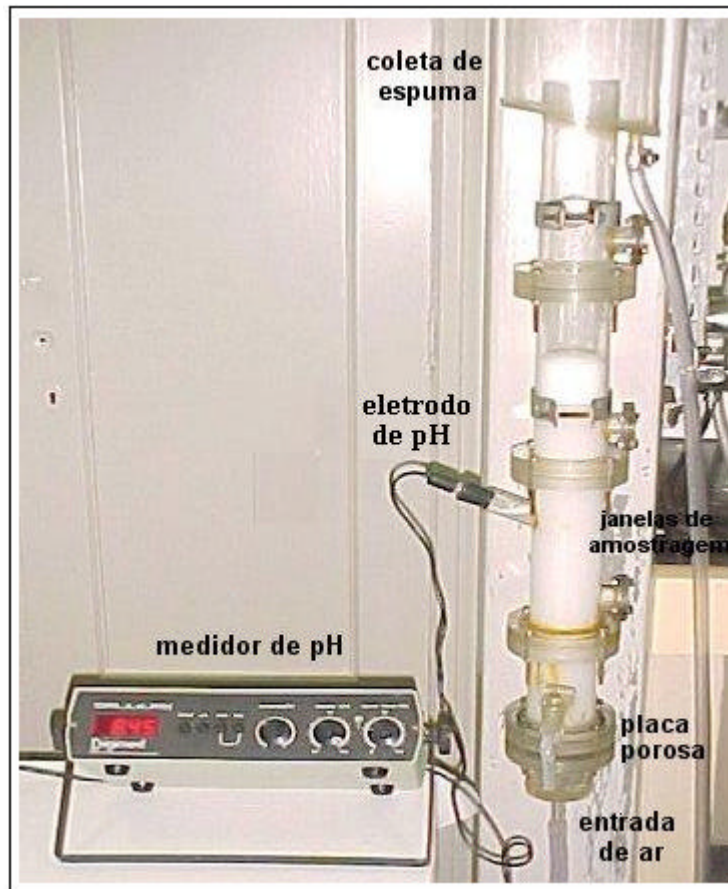


## 10.2. Curva de Calibração do Rotâmetro





### 10.3. Linha de Montagem da Coluna de Flotação



**10.4.****Relação entre a remoção de cromo (III) e a concentração de coletor**

Tempo (min)	$0,5 \times 10^{-4} M$	$1 \times 10^{-4} M$	$1,5 \times 10^{-4} M$
0	0.00	0.00	0.00
5	62.36	85.87	40.01
10	85.92	91.56	79.97
15	92.23	92.51	91.45
20	94.32	92.89	94.47
30	95.62	93.75	95.75
50	95.57	96.20	96.07

**10.5.****Relação entre a remoção de cromo (III) e a vazão de ar**

Tempo (min)	$1,41 \text{ mL.s}^{-1}$	$2,10 \text{ mL.s}^{-1}$	$3,46 \text{ mL.s}^{-1}$
0	0.00	0.00	0.00
5	86.73	85.87	83.03
10	91.50	91.56	92.11
15	92.01	92.51	94.22
20	93.11	92.89	94.97
30	93.11	93.75	95.62
50	94.22	96.20	95.35

**10.6.****Relação entre a remoção de cromo (III) e a concentração de espumante**

Tempo (min)	Sem	0.10%	0.20%	0.50%
0	0.00	0.00	0.00	0.00
5	35.09	85.87	76.78	54.30
10	62.63	91.56	79.56	66.63
15	75.19	92.51	83.62	82.20
20	77.58	92.89	86.24	87.00
30	79.21	93.75	88.14	79.85
50	81.21	96.20	86.39	93.86

## 10.7.

**Varição do Potencial zeta do hidróxido de cromo em função do pH na presença de KCl**

pH	Potencial Zeta (mV)		
	KCl $10^{-2}$ M	KCl $10^{-3}$ M	KCl $10^{-4}$ M
6.4	25.1		
7.13	27.4		
8.11	18.9		
8.22	14.6		
8.8	-10.3		
9.02	-15.6		
10.02	-33.6		
11.02	-33.1		
6.01		30.2	
7.02		24.2	
8.12		19.3	
8.74		-13.5	
9.05		-25.9	
10.02		-33.7	
11.02		-35.7	
6.39			30.3
7			27.3
8.25			14.5
8.94			-19.6
10.05			-34.2
11.04			-37.1

## 10.8.

Efeito da adição de coletor (DSS) no potencial zeta do hidróxido de cromo

pH	Potencial Zeta (mV)			
	Sem DSS	10 <sup>-5</sup> M DSS	10 <sup>-4</sup> M DSS	10 <sup>-3</sup> M DSS
6	42.8			
7	33.3			
8	14.9			
9	-29			
10	-39.1			
11	-40			
6		22.2		
7		0.5		
8		-13		
9		-32.6		
10		-40.9		
11		-43.8		
6			2.8	
7			-2.95	
8			-14.3	
9			-31.7	
10			-41.3	
11			-41.4	
6				-3
7				-3.2
8				-15
9				-26.9
10				-35.9
11				-38.1

## 10.9.

Efeito da adição de íons cromo (III) no potencial zeta do  $\text{Cr}(\text{OH})_3$ 

pH	Potencial Zeta (mV)			
	Sem $\text{Cr}^{3+}$	$10^{-5} \text{M Cr}^{3+}$	$10^{-4} \text{M Cr}^{3+}$	$10^{-3} \text{M Cr}^{3+}$
6	42.8			
7	33.3			
8	14.9			
9	-29			
10	-39.1			
11	-40			
6		17.4		
7		8		
8		-2		
9		-31		
10		-38.3		
11		-39.5		
6			26.7	
7			18.7	
8			15.9	
9			-23.9	
10			-38.1	
11			-39.7	
6				31.2
7				29.3
8				23.8
9				1.6
10				-27.3
11				-37.7