

## Referências Bibliográficas

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## Apêndice A

### A.1. Cálculo de $C_R$

A dedução do valor de  $C_R$  da eq. (5.1) é mostrada a seguir:

$$\begin{aligned}
 C_R &= \left( \frac{gC}{100 \text{ gTTNTs}} \cdot \frac{1 \text{ mol } N}{14g \text{ N}} \cdot \frac{6,022^{23} \text{ átomos } N}{1 \text{ mol } N} \cdot \frac{1 \text{ molécula APTES enxerta}}{1 \text{ átomos } N} \right) \cdot \\
 &\quad \left( \frac{1 \text{ mol APTES enxerta}}{6,022^{23} \text{ moléculas APTES enxerta}} \cdot \frac{1000 \text{ mmol APTES enxerta}}{1 \text{ mol APTES enxerta}} \right) \\
 C_R &= \frac{10N}{14n}
 \end{aligned}$$

### A.2. Cálculo de $C_T$

O valor de  $C_T$  para TTNTs/L foi calculado com a seguinte expressão:

$$\begin{aligned}
 C_T &= \left( \frac{5,8 \text{ grupos - OH}}{\text{nm}^2} \cdot \frac{1 \text{ nm}^2}{(10^{-9})^2 \text{ m}^2} \cdot \frac{283,1027 \text{ m}^2}{1 \text{ gTTNT}} \right) \cdot \\
 &\quad \left( \frac{1 \text{ molécula APTES}}{1 \text{ grupo - OH}} \cdot \frac{1 \text{ mol APTES}}{6,022 \cdot 10^{23} \text{ moléculas APTES}} \cdot \frac{1000 \text{ mmol APTES}}{1 \text{ mol APTES}} \right) \\
 C_T &= 2,73 \frac{\text{mmol APTES}}{\text{gTTNTs}}
 \end{aligned}$$

O mesmo cálculo foi realizado para obter o valor de  $C_T$  dos TTNTs/H, modificando unicamente o valor da área específica para  $283.1027 \text{ m}^2/\text{g}$ .