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A

Apêndice

A.1 Efeito de K_1

Nas figuras A.1 e A.2 evidenciamos o efeito do valor da constante de equilíbrio K_1 no inchamento teórico calculado para as concentrações de NaCl 1 M e 0,01 M.

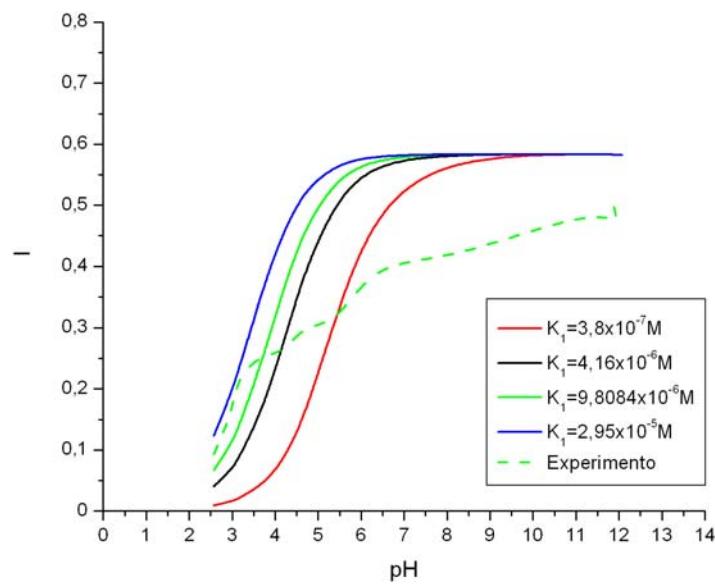


Figura A.1: Efeito do valor de K_1 no Inchamento. NaCl 1 M, $E_s=2,5$ MPa, $K_2=0,8482$ M $^{-1}$, $\delta=1\times10^{-9}$ m, $As=1600$ m $^2/g$.

Em ambas curvas notamos que o aumento de K_1 conduze a maiores valores de inchamento na faixa de $pH \approx 2$ a $pH \approx 7$. Sendo K_1 a constante de equilíbrio que favorece as reações desprotonação, tal aumento de inchamento é associado a um aumento da densidade de carga superficial nessa faixa de pH onde são observados os maiores inchamento relativos no experimento (ver figura 6.3).

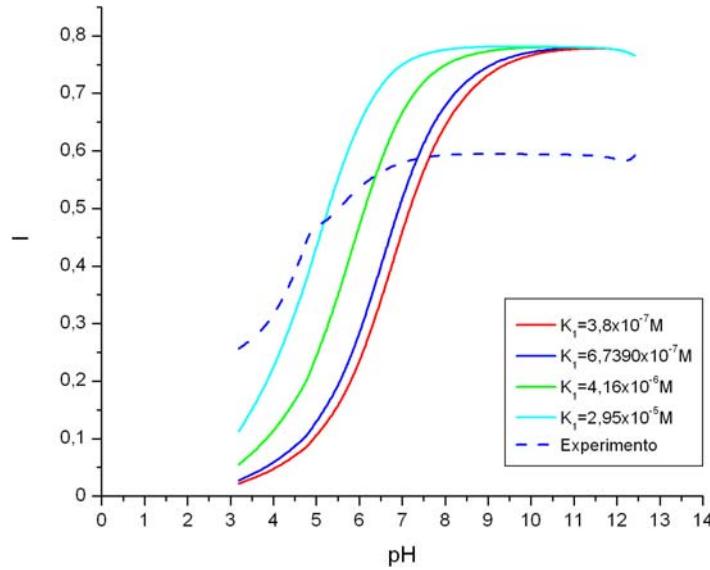


Figura A.2: Efeito do valor de K_1 no Inchamento. NaCl 0,01 M, $E_s=2,5 \text{ MPa}$, $K_2=0,6987 \text{ M}^{-1}$, $\delta=1\times 10^{-9} \text{ m}$, $As=1600 \text{ m}^2/\text{g}$.

A.2 Efeito de K_2

Nas Figuras A.3 e A.4 apresentamos o efeito da mudança de K_2 sobre as curvas de inchamento teórico para NaCl 1 M e 0,01 M. Em ambas curvas observamos que ao contrário de K_1 , o incremento de K_2 ocasiona a diminuição do inchamento teórico. Associamos tais resultados ao aumento da sorção de sódio nos sítios ativos da resina. Tal fenômeno é traduzido na diminuição da densidade de carga superficial e consequentemente em menores valores de inchamento.

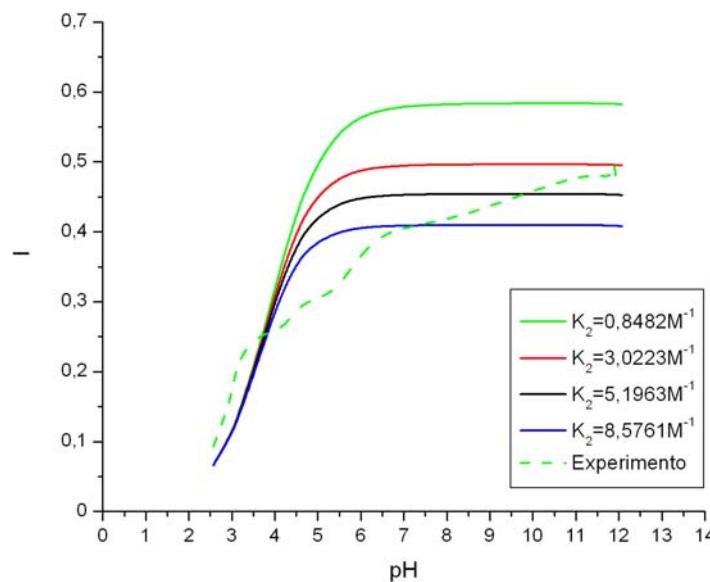


Figura A.3: Efeito do valor de K_2 no Inchamento. NaCl 1 M, $E_s=2,5 \text{ MPa}$, $K_1=6,739\times 10^{-7} \text{ M}$, $\delta=1\times 10^{-9} \text{ m}$, $As=1600 \text{ m}^2/\text{g}$.

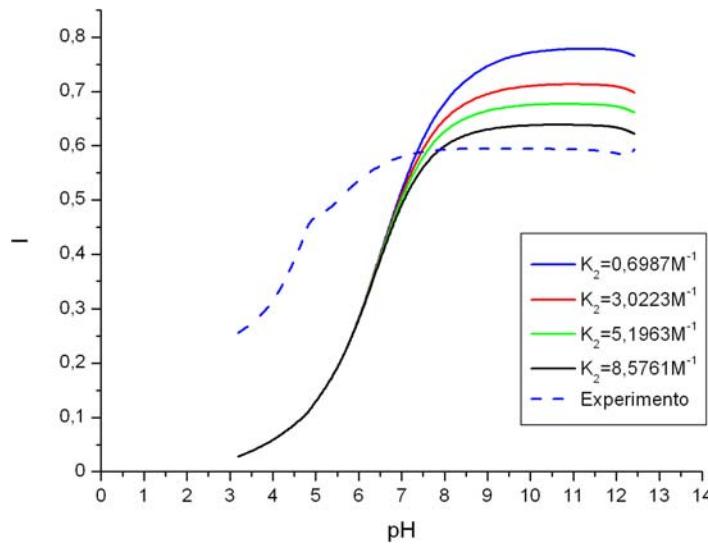


Figura A.4: Efeito do valor de K_2 no Inchamento. $\text{NaCl} 0,01 \text{ M}$, $E_s=2,5 \text{ MPa}$, $K_1=9,8084 \times 10^{-6} \text{ M}$, $\delta=1 \times 10^{-9} \text{ m}$, $As=1600 \text{ m}^2/\text{g}$.

A.3 Efeito de E_s

Nas figuras A.5 e A.6 apresentamos o efeito do aumento do módulo de Young sobre o inchamento teórico para as concentrações de NaCl 1 M e 0,01 M. Em ambos casos observamos que o inchamento diminui com o aumento do módulo de Young. O resultado típico de matérias modelados mediante a lei de Hook é justificado, devido a que o aumento força elástica é traduzido na diminuição do espaçamento entre as placas H para satisfazer o balanço de forças dado pela equação (6-10a).

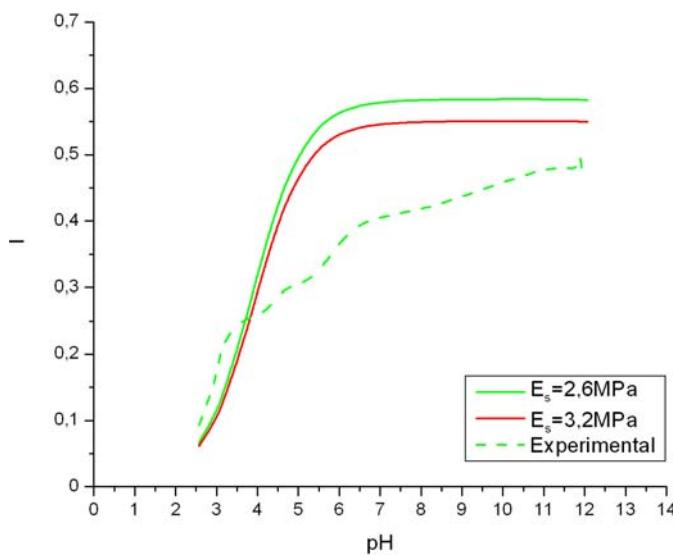


Figura A.5: Efeito do valor do módulo de Young no Inchamento - NaCl 1 M. Valores de K_1 e K_2 obtidos por HYPERQUAD, $\delta=1 \times 10^{-9} \text{ m}$, $As=1600 \text{ m}^2/\text{g}$.

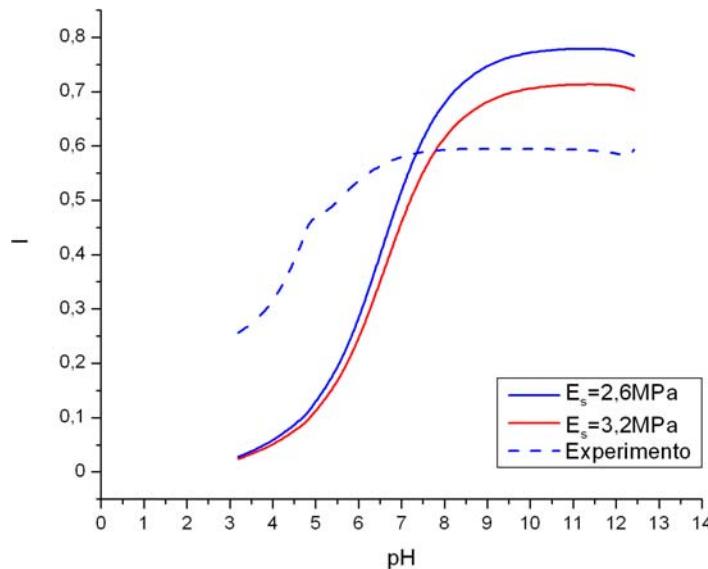


Figura A.6: Efeito do valor do módulo de Young no Inchamento - NaCl 0,01 M. Valores de K_1 e K_2 obtidos por HYPERQUAD, $\delta=1\times10^{-9}$ m, $As=1600$ m^2/g .

A.4

Efeito de As

Nas figuras A.7 e A.8 apresentamos os inchamentos teóricos em função do pH para 1 M e 0,01 M obtidos para dois valores de As . Fisicamente a diminuição do valor da área superficial pode ser descrita como se a carga total da resina for re-distribuída em uma área menor. A maior proximidade entre os grupos ativos carregados na superfície, produz maiores repulsões eletrostáticas entre eles, assim como maiores interações com as espécies iônicas na camada difusa, o que é traduzido em ambas curvas no aumento dos valores de inchamento.

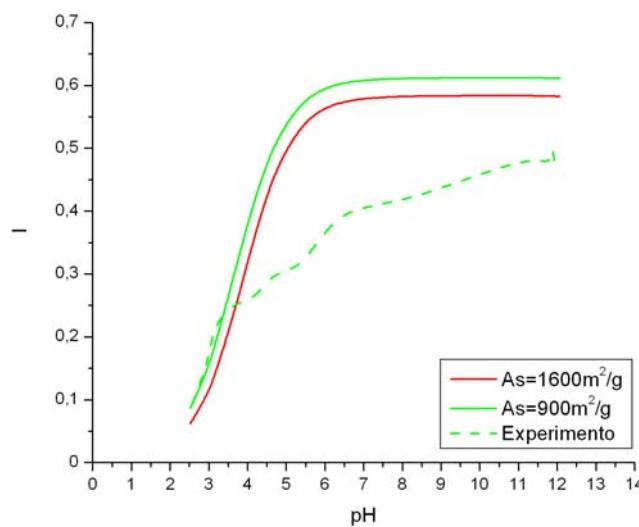


Figura A.7: Efeito do valor da área superficial no Inchamento - NaCl 1 M. Valores de K_1 e K_2 obtidos por HYPERQUAD, $E_s=3,2$ MPa, $\delta=1\times10^{-9}$ m.

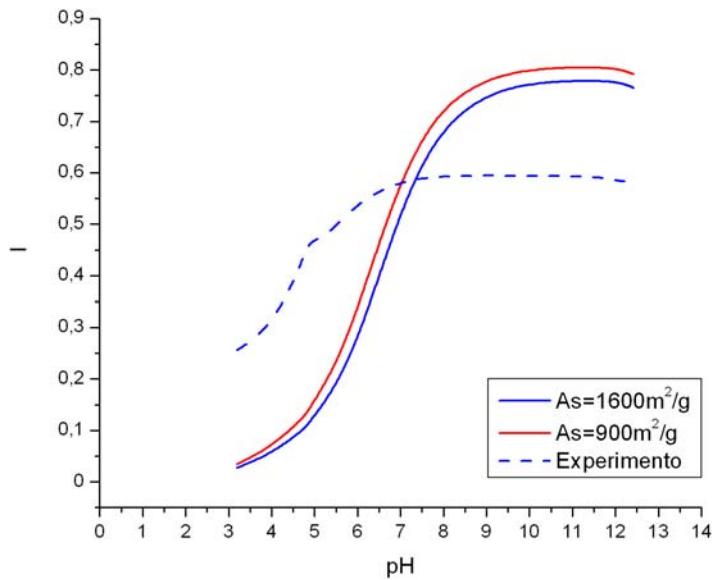


Figura A.8: Efeito do valor da área superficial no Inchamento - NaCl 0,01 M. Valores de K_1 e K_2 obtidos por HYPERQUAD, $E_s=3,2$ MPa, $\delta=1\times 10^{-9}$ m.

A.5 Efeito de δ

Nas figuras A.9 e A.11 apresentamos as curvas de H em função do pH construídas com dois valores de δ para as concentrações de NaCl 1 M e 0,01 M. Como observado nos resultados numéricos das figuras A.9 e A.11 a diminuição do δ é compensada através da diminuição do espaçamento entre as placas para satisfazer o balanço de forças (equação 6-10a). Devido à ordem de grandeza dos valores de H e δ , o inchamento calculado através da equação (6-10b) será maior para o menor valor de δ , como mostrado nas figuras A.10 e A.12.

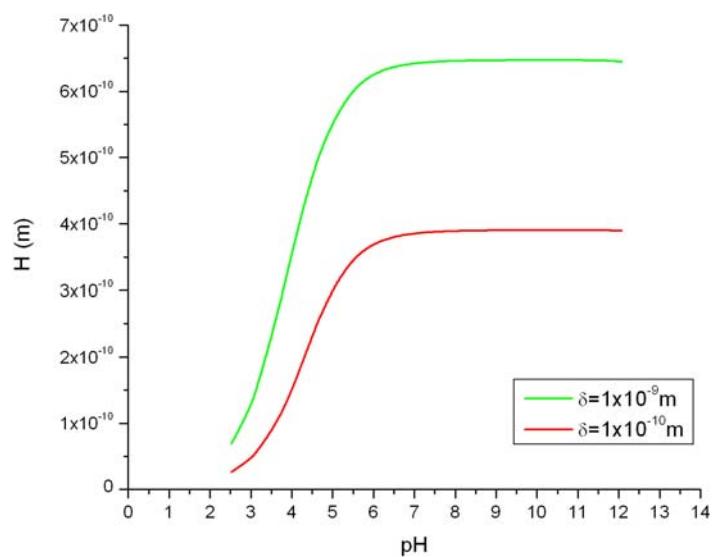


Figura A.9: Efeito do valor da espessura da placa no H - NaCl 1 M. Valores de K_1 e K_2 obtidos por HYPERQUAD, $E_s=3,2$ MPa, $As=900 \text{ m}^2/\text{g}$.

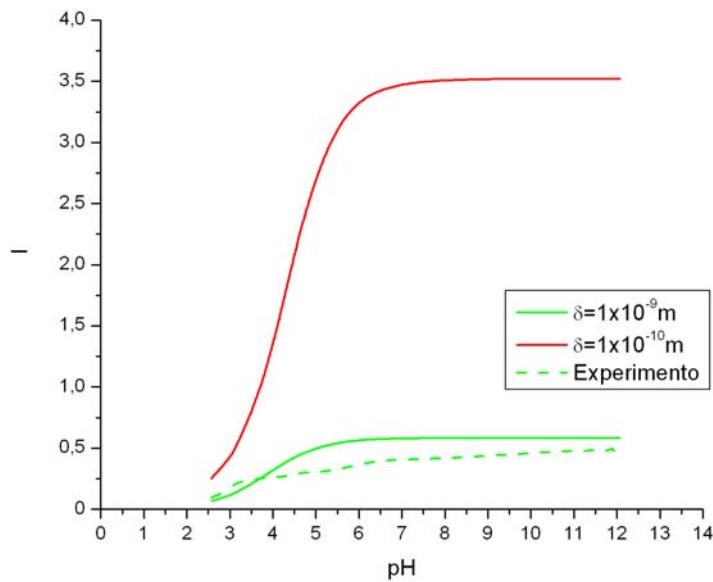


Figura A.10: Efeito do valor da espessura da placa no Inchamento - NaCl 1 M. Valores de K_1 e K_2 obtidos por HYPERQUAD, $E_s=3,2 \text{ MPa}$, $As=900 \text{ m}^2/\text{g}$.

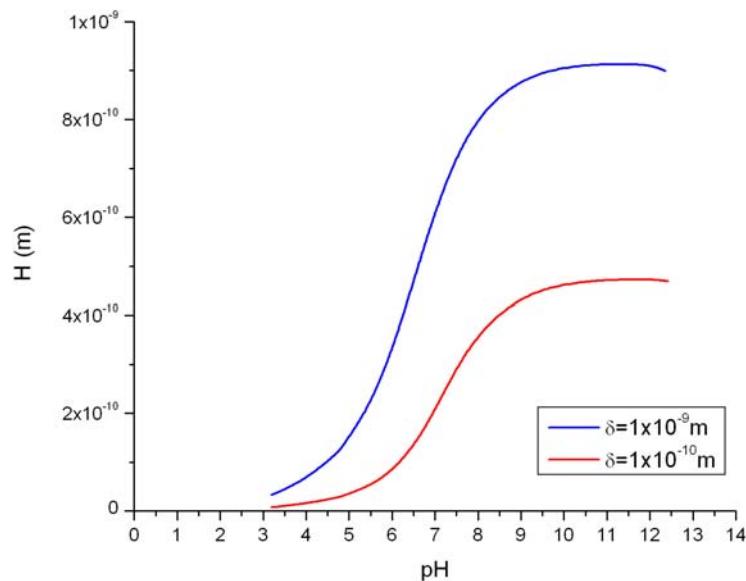


Figura A.11: Efeito do valor da espessura da placa no H - NaCl 0,01 M. Valores de K_1 e K_2 obtidos por HYPERQUAD, $E_s=3,2 \text{ MPa}$, $As=900 \text{ m}^2/\text{g}$.

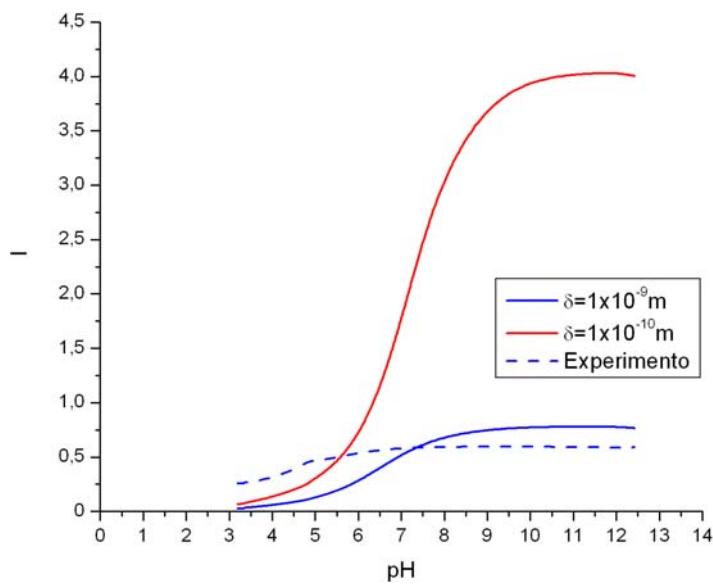


Figura A.12: Efeito do valor da espessura da placa no Inchamento - NaCl 0,01 M. Valores de K_1 e K_2 obtidos por HYPERQUAD, $E_s=3,2 \text{ MPa}$, $As=900 \text{ m}^2/\text{g}$.